3. LITHOLOGY AND CLAY MINERALOGY OF SEDIMENTS FROM SITE 337, DSDP LEG 38

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DESCRIPTIONS OF THE SEDIMENT SERIES

The sediments cored at Site 337 to the depth of 180 meters are divided according to their lithologic and mineralogic characteristics into two series (Figure 1).

Series 1 (0-42.2 m, Samples 2061-2081)

Pelitomorphous clays with foraminiferal shells, coccolith marls, unsorted sand-silt-clays, rare tuff layers, and an absence of siliceous organisms. The series is divided into two subseries.

Subseries 1 (0-28 m, Samples 2061-2076)

Alternating slightly silty marls and foraminiferal clays with polymictic, unsorted sandy/clayey sediments with terrigenous material which includes: fragments of basalts, granites, hornblende and biotite schists, and quartzites. These were derived via ice-rafting.

Clay Mineralogy

Sediments of Subseries 1 (Pleistocene) contain in the clay fraction: illite, chlorite, montmorillonite, and a small amount of kaolinite (Table 1). The principal clay component is illite. The illite does not have expandable interlayers in its structure; the octahedral positions are predominantly populated by Al cations. This is a polytype 1 Md modification, sometimes with an admixture of 1 M. The illite polytype was determined by the electron diffraction method of oblique structures, or with the aid of diffraction patterns of unoriented preparations preheated to 550°C. The illite content in the subseries is constant and averages about 60%.

A second clay component is chlorite with a maximum content of 30% or less. It belongs to the trioctahedral modification. Intensity ratios of the basal reflections (a low intensity of 001 and 003 reflections) indicate a heightened content of Fe cations in the octahedral structure, i.e., it is an Fe-chlorite. However, the mineral has a very low resistance to thermal treatment. After a sample is heated for 1 hr at 550°C, all reflections characteristic of chlorite disappear, or there are weak reflections at d = 13.2-13.6Å. This indicates a defective structure and, specifically, an apparently "insular" (partly hydrated) structure of one-level brucite layers. The data obtained suggest that, to a large extent, the chlorite was formed as a result of structural alterations of biotite. The content of biotite in the sediments is high.

In this subseries, the montmorillonite content is insignificant, averaging 15%-20% with a maximum of 30%. Another common clay mineral (Samples 2062, 2063, 2067, 2073) is a mixed-layer clay with the following diffraction characteristics: the diffraction pattern of the untreated sample shows a broad continuous reflection, appearing as a "plateau," in the region of d = 11-

14Å; in the glycerine-saturated sample, the diffraction pattern shows a broad diffuse peak from 14 to 19Å with a maximum at $d \approx 17-18$ Å. Also, the reflection at d \approx 4.7Å becomes broader. The data indicate the mixedlayer clay has montmorillonite layers with varying degrees of hydration alternating in its structure in an unordered fashion and in varying proportions. The degree of hydration, for the most part, is dependent upon the type of exchange cations. The cations seem, for the most part, to be K, Ca, and Mg. After glycerine saturation, the montmorillonite layers expand to 17.7Å. However, the vermiculite interlayers which are saturated with K cations, expand to only 14Å. Thus, it appears that after glycerine/saturation a mixed-layer phase of irregularly alternating 17.7 and 14Å layers is formed, with a prevalence of the 17.7Å layer. After heating, all layers contract to 9.6-9.8Å. It is possible that the mineral is also a product of biotite disintegration. The transformation of biotite through vermiculite to dioctahedral montmorillonites is a well-known process and is a widespread occurrence in sedimentary rocks and weathering crusts.

Another modification of montmorillonite (Samples 2065, 2066, 2070) is a mineral whose interlayers contain primarily K cations in the exchange state. Therefore, the diffraction pattern of an untreated sample has the montmorillonite reflection with d = 11Å. This is often seen only as a background of heightened intensity, a "train" spreading out from the illite reflection (001) towards lower angles. However, the mineral retains its capacity to expand after saturation. The montmorillonite contains iron and dissolves in 1N HCl. Two alternative explanations for the structural characteristics of this mineral are suggested: the montmorillonite is authigenic and developed from transformation of an alkaline pyroclastic mineral. Potassium accumulated either as a result of adsorption of these cations from seawater, or the K cations could have been supplied from the disintegration of basalt or basalt glass. If it is so, the process can be considered an "embryonic" form of glauconitization; however, it is more likely that the montmorillonites were also formed by the alteration of biotite. A portion of the K cations are removed from interlayers. The remaining ones cannot bond the silicate layers, and thus the layers become capable of expanding.

Subseries 2 (28-42.2 m, Samples 2077-2081)

A predominance of coccolith marls. A thin ash layer. No ice-rafted clastic material.

Clay Mineralogy

The clay fraction of Subseries 1 consists almost entirely of montmorillonite (\approx 70%), illite (\approx 20%-25%), and kaolinite (5%-10%) (Table 1). A characteristic

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AGE	DEPTH SERIES SUBSERIES, PACKETS	LITHOLOGIC COLUMN	CORE, SECTION, INTERVAL (cm) AND SAMPLE NO. (GIN)	THIC	KNESS OF SI UBSERIES (r	ERIES n)	LITHOLOGIC DESCRIPTION OF SERIES
PLEISTOCENE			1-3, 86-88 (2063) 1-4, 87-89 (2064) 1-5, 85-87 (2065; 1-6, 109-111 (2066) 2-1, 90-92 (2067) 2-2, 90-92 (2068) 2-3, 90-92 (2070) 2-4, 90-92 (2070) 2-5, 100-102 (2071) 2-6, 90-92 (2072) 3-7, 30-32 (2073)		42.2	28	Pelitomorphous clays with foraminifera, coccolith marls, unsorted silt-clay.
PLIOCENE	28.0		3-6, 8-10 (2075) 4-1, 90-94 (2076) 4-4, 91-93 (2078) 5-1, 68-70 (2079) 5-2, 100-102 (2080) 5-4, 100-102 (2081)			14.2	Rare thin tuff layers; complete absence of siliceous organisms.
UNDATED	66.0		6-2, 67-69 (344) 7-1, 75-77 (345) 7-2, 75-77 (346) 7-3, 75-77 (347) 7-4, 77-79 (348) 7-5, 77-79 (349) 7-6, 77-79 (350) 8-1, 75-77 (351) 8-2, 76-78 (352)		65	18	Pelagic clays of both zeolitic and sub- stantially siliceous types with individual thin layers of sandy/silty clays and vol- canic glasses; a characteristic feature is a complete absence of calcareous organisms and a widespread distribution of siliceous organisms.
EARLY OR MIDDLE OLIGOCENE	104.0		9-2, 77-79 (353) 9-3, 75-77 (354) 9-4, 75-77 (355) 9-5, 75-77 (356) 9-6, 75-77 (356) 10-2, 75-77 (358) 10-3, 75-77 (359) 10-4, 75-77 (360) 10-5, 75-77 (361) 10-6, 75-77 (363) 11-2, 75-77 (364) 11-3, 75-77 (364) 11-3, 75-77 (365) 11-4, 78-80 (366) 11-5, 79-84 (367)		(70.8)	38	
		©-©-	12-2, 76-80 (369) 12-3, 76-78 (370) 12-5, 119-121 (371)			9	
							ALTERED BASALTS

Figure 1. Stratigraphic column at Site 337, including lithologic descriptions and descriptions of clay minerals and clay mineral associations.

LITHOLOGIC DESCRIPTION OF SUBSERIES	DESCRIPTION OF CLAY AND AUTHIGENIC MINERALS	CLAY MINERALS
Alternating low-silt marls and foraminifera clays with polymictic unsorted sand-clay; terrigenous material consists of fragments of basalts, granites, hornblende, and biotite schists, quartzites. For the most part associated with ice-rafting.	Dioctahedral hydromica (illite), more often without expandable layers, modifications IMd and IM 60%; trioctahedral chlorite with de- fective brucite layer (20-30%); K-montmoril- lonite (15-20%); med-layer products of transformation of biotite of the chlorite- vermiculite series; admixture of kaclinite (5-10%).	Hydromicaceous with chlor- ite and montmorillonite; abundant products of stadial transformation of biotite.
Predominance of coccolith marls; a thin ash layer; absence of clastic or ice-rafted debris.	Na, Ca, Mg-montmorillonite, Al, Fe, or Al-Fe containing (70%); illite (more often without expandable layers) (20-25%); de- fective chlorite and kaolinite make up a small admixture in individual thin layers.	Montmorillonite with illite.
Montmorilionite clays with thin layers of basic tuffs and ashes and some fragments of volcanic glasses in clay; a phillipsite association is present in tuffs.	Well-crystallized Ca-montmorillonites combined with mixed-layer montmorillonite- hydromicas forming a spectrum with ratios from 0 to 100%; presence of phillipsite.	Mixed-layer, montmoril- lonitic association with phillipsite.
Montmorillonite clays with abundant remains of siliceous organisms (Radiolaria, diatoms, spicules), increasing down section; frag- ments of basic volcanic glasses are immersed in clay; a characteristic feature is the development of montmorillonite with abnormally high spacings of up to 19.6Å (in saturated state) at the expense of SiO ₂ absorption.	Poorly cyrstallized montmorillonites with abnormally high values of d(001) and a broad profile of basal peaks linked up with sorption of amorphous phases of SiO; ad- mixtures of hydromica, chlorite, and kaolinite.	Silica-montmoril- lonite association.
Montmorillonite clays impoverished in siliceous organisms and enriched in cry- stallized volcanic glasses; breccia-like fragments of altered basalt.	Montmorillonite with mixed-layer hydro- mica-montmorillonite series; presence of clinoptilolite.	Mixed-layered mont- morillonite association with clinoptilolite.
ALTERED BASALTS		

Figure 1. (Continued).

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 TABLE 1

 Data From X-Ray Analysis of Clay and Other Minerals, Site 337

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			of								
Age	Depth (m)	Core, Section, Interval (cm) and (GIN) Sample No.	Frac- tion (mk)	Montmorillonite	Mixed-Layered Minerals ^a	Hydromica	Chlorite	Kaoli- nite	Zeolites	Other Minerals	Clay Mineral Associations
		1-3, 86-88(2063)	<1	Montmorillonite (little)	(M-V) product of changing biotite, closely related to montmorillonite	Illite ~ 60%	Trioctahedral vari- ety 15-20%	10-20%		Quartz	
		1-4, 87-89(2064)	<1	Montmorillonite (little)	7 — 1	Illite ~ 60%	Trioctahedral vari- ety 15-20%	10-20%		Quartz	
Pleistocene		1-5, 85-87(2065)	<1	Montmorillonite (little); contains micaceous unlabile packets		Illite without labile beds ~ 60%	Trioctahedral, de- veloped on biotite with an imperfect brucite bed	10-20%		Quartz	
		1-6, 109-111(2066)	<1	K-montmorillonite		Illite without labile beds ~ 60%	Trioctahedral, de- veloped on biotite with an imperfect brucite bed	10-20%			
		2-1, 90-92(2067)	<1		(M-V) product of transformation of biotite (soluble in HC1)	Illite ~ 60%	10-20%	10-20%			
		2-2, 90-92	<1		(M-V) product of transformation of biotite (soluble in HC1)	Illite 1M and 1 Md	10-20%	10-20%			Hydromicaceous with chlorite and montmorillonite; abundance of products of store
		2-3, 90-92(2069)	<1	755	(M-V) product of transformation of biotite (soluble in HC1)	Illite ~ 60%	10-20%	10-20%			transformation of biotite
		2-4, 90-92(2070)	<1	K-montmorillonite	_	Illite ~ 50% (with- out labile beds)	Trioctahedral vari- ety 30% with an im- perfect brucite bed; does not bear heat-	5-10%			
		2-5, 100-102(2071)	<1	K-montmorillonite	-	-	Trioctahedral, more perfect structure, can be preserved after heating at 550°C				
		2-6, 90-92(2072)	<1	Ca, K-montmorillo- nite up to 30%		Illite $\sim 50\%$ with good crystallization	Trioctahedral vari- ety 20-30%	to 5%		Quartz, feldspars	
		3-1, 30-32(2073)	<1	Montmorillonite (little)	(M-V) intermediate phase of transforma- tion of biotite	Illite 50-60%	Trioctahedral vari- ety after biotite	to 10%		Quartz, feldspars	
		3-6, 8-10(2075)	<1	K, Na-montmorillo- nite aluminic, ferru- ginous up to 30%	—	Illite ~ 50%	Trioctahedral vari- ety with an imper- fect brucite bed	to 5%			
		4-1, 90-94(2076)	<1	Na, Ca, Mg-montmoril- lonite ~ 50%	<i></i>	lllite ~ 40%	Trioctahedral vari- ety with an imper- fect brucite bed	5-10%			
		4-4, 91-93(2078)	<1			Illite 20-30%	-	5-10%			Montmorillonitic

with illite

		5-1, 68-70(2079)	<1	Montmorillonite aluminic up to 70%		Illite 20-30%		5-10%				
		5-2, 100-102(2080)	<1	Ca, Mg-montmorillo- nite slightly ferrugi- nous ~ 90%	-	Illite 20-30%	-		Abundance of zeolite on volcanic glass			
Older(?)		5-4, 100-102(2081)	<1	Mg, Ca-montmorillo- nite ferruginous ~90%	5-	_	Traces					
cene and	50.2	6-2, 67-69(344)	Rock	Calcic form 50-60%	Spectrum from 0 to 100%, 15-20%	20% labile interbeds of montmorillontic type $\sim 15\%$		-	Phillipsite 10-15%	Quartz, feldspars		
Plio	58.7	7-3, 75-77(346)	Rock	Calcic form 60-70%	Spectrum from 0 to 100% ~ 15%	20% labile interbeds of montmorillonitic type $\sim 10\%$		-	27.).	Quartz, feldspars, amorph. SiO ₂	Mixed-layer montmorillonitic with phillipsite	
	61.8	7-4, 80-82(348)	<10	Calcic form 75-80%	20% interbeds of hydromicaceous type $\sim 10\%$	Illitic type ~ 10%	-	5%		Quartz		
	63.3	7-5, 77-79(349)	<10	Magnesium-calcic form ~ 80%	Not over 69%	Illitic type $\sim 15\%$	-	-	-	Quartz		
	68.3	8-2, 76-78(352)	Rock	Sodium-magnesium- calcic form $\sim 80\%$	Spectrum from 0 to 100%, 15-20%	20% labile interbeds of montmorillonitic type $\sim 10\%$		3—	-	Quartz, feldspars, amorph. SiO ₂		
	79.3	9-3, 75-77(354)	<10	Calcic form ~80%	15-20% interbeds of micaceous type ~ 10%	Illitic type ~ 15%	-	Little	Ξ	Quartz, feldspars		
	82.3	9-5, 75-77(356)	<10	Calcic form ~80%	Spectrum with 20% interbeds of mica- ceous type $\sim 10\%$	Illitic type ~ 10%	-	Little	-	Quartz, feldspars	Siliceous montmorillonitic	
sne	87.3	10, 275-77(358)	<10	Calcic, poorly crys- tallized form with adsorbed $SiO_2 \sim 50\%$		Up to 10% of labile interbeds of mont- morillonitic type, 30%	10%	10%		Quartz, amorph. SiO ₂		
ddle Oligoce	90.3	10-4, 75-77(360)	<10	Poorly crystallized with adsorbed $SiO_2 \sim 60\%$	-	Up to 10% labile interbeds of mont- morillonitic type, 30%	5-10%	5-10%	-	Quartz, feldspar amorph. SiO ₂		
dy or Mi	93.3	10-6, 75-77(362)	<10	Poorly crystallized with adsorbed SiO ₂ ~60%	-	Illitic type ~30%	10%	10%	3 1	Quartz, feldspars		
Eau	95.3	11-1, 75-77(363)	<10	Poorly crystallized with adsorbed SiO ₂ ~75%	8 <u>0</u>	Illitic type ~15%	10%	Little		Quartz, feldspars		
	96.8	11-2, 75-77(364)	Rock	Poorly crystallized with adsorbed $SiO_2 \sim 75\%$		Up to 20% labile interbeds of mont- morillonitic type 10-15%	5%	Little	-	Quartz, amorph. SiO ₂		
	99.8	11-4, 78-80(366)	<10	Poorly crystallized with adsorbed SiO ₂ ~60%	0	Illitic type 30%	10%	Little		Quartz, feldspars, amorph. SiO ₂		

25

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	Clay Mineral Associations			Mixed-layer montmorillonitic with clinoptilolite		
	Other Minerals	Quartz, amorph. SiO2	Quartz, amorph. SiO2	Quartz, amorph. SiO2 Ca Mg-carbon- ate	Quartz, amorph. SiO ₂ , Ca- Mg-carbonate	Quartz, Ca- Mg-carbonate
	Zeolites	a	Clinoptilo- lite(?)	Clinoptilo- lite(?)	Clinoptilo- lite	Clinoptilo- lite
	Kaoli- nite	Little	Little	Little	1	Ū
	Chlorite	10%	10%	5%	ì	Ĩ.
Continued	Hydromica	Illitic type 30%	Illitic type 20%	Illitic type 10%	Up to 20% labile interbeds of mont- morillonitic type, ~15%	Up to 10% labile interbeds of mont- morillonitic type ~10%
	Mixed-Layered Minerals ^a	Ţ	ī	Ť	Spectrum from 0 to 100% ~10%	Spectrum from 0 to 100% ~5%
	Montmorillonite	Poorly crystallized with adsorbed SiO ₂ ~60%	Poorly crystallized with adsorbed SiO ₂ ~70%	Calcic form 80%	Calcic form 75-80%	Calcic form ~85%
	Size of Frac- tion (mk)	<10	<10	<10	Rock	<10
	Core, Section, Interval (cm) and (GIN) Sample No.	11-5, 79-81(367)	12-1, 78-80(368)	12-2, 76-80(369)	12-3, 76-78(370)	12-5, 119-121(371)
	Depth (m)	101.3	104.8	106.3	107.8	111.2
	Age		əuə	ogilO slbbiM	Early or	

feature of the montmorillonites is a varying iron content in its structure. This mineral characteristic was qualitatively determined by the distribution of the basal reflections in the diffraction patterns of glycerinesaturated, and 550°C preheated samples. Diffraction patterns of strongly ferruginous montmorillonites, when saturated with glycerine, have a relatively low intensity of reflections ([004], [005] and [006]). Thus, they differ from the diffraction patterns of aluminum montmorillonites. Diffraction patterns of preheated samples give particularly clear data on the extent of the montmorillonite iron content. Strongly ferruginous montmorillonites have no reflection at $d(002) \cong 4.7-4.8$ Å. In contrast, dehydrated aluminum varieties have a rather intense (002) reflection. An intermediate intensity corresponds to an intermediate iron content for the montmorillonite. Chemical analyses were available for some samples, which enabled a more reliable division of the montmorillonites according to their iron content (Table 2). They were divided into three groups: predominantly aluminum, aluminum-iron, and iron-containing.

Analysis of the data has shown that, in this subseries, montmorillonites of varying iron contents alternate. There is a relative increase in iron content (Samples 2079, 2082, 2080, 2078, 2081). The mica-like component of the subseries belongs either to nonexpanding illites, or to hydromicas. The hydromicas usually contain approximately 20% expandable interlayers. Chlorite, present in the subseries, is also trioctahedral with a defective brucite layer.

Series 2 (Cores 5-4 to 12-5, 42.4-113.0 m; early or middle Oligocene and Pliocene)

The series consists of a yellow, brown, gray, yellowish-orange clay, with rare, indistinct layers of sandy siltstones, volcaniclastic tuffs, and ashes. A zeolite mineralization is confined to the upper portion. The bulk of the sediments contains remains of siliceous organisms, which first appear in the upper part of the series, and are unvaryingly present to the base.

The clay, forming the bulk of sediments, is characteristic. It consists of Fe-montmorillonites (typical minerals of pelagic sediments) developed at the expense of the transformation (halmyrolysis) of tholeiitic basalts (Table 2, Samples 344, 346). Characteristics of this mineral group and their origin have been studied in detail on materials from DSDP Legs 2, 20, 27 (Kossovskaya and Shutov, 1975; Kossovskaya et al., 1963, 1975).

A distinctive montmorillonite was noted in Sample 344, containing 18.56% Fe₂O₃. This clay was formed by the accumulation of decomposed hyaloclastics, and a hydrothermal influx of ferruginous solutions. Another characteristic of this sample is the relatively high manganese content (to 0.28%). The remaining samples of the entire series fall within the composition range of pelagic clay associated with the decay of basalt hyaloclastics. However, with increasing depth in the series, there is a slight increase in Al₂O₂ (15.63% to 16.70%), a drop in Fe₂O₃ (10.02% to 6.81%), and an increase in K₂O (1.92% to 2.66%) and Na₂O (3.73% to 4.09%).

						Bulk F	ractions					
	<10	P		Series 2								
	<10 µm	Fraction		E	arly or M	idale Oli	gocene an	a Phocene				
	Subseries 1 2072	Subseries 2 2077	344	Subse 344	eries 2 346	346	Subs 352	eries 2 352	Subse 370	370 370		
SiO ₂	51.75	49.87	41.14	38.51	48.00	44.02	47.58	42.70	47.96	43.10		
TiO ₂	0.98	1.20	1.34	1.41	1.05	1.13	1.22	1.35	1.12	1.23		
Al203	17.30	15.38	12.11	12.73	14.39	15.63	14.70	16.27	15.15	16.70		
Fe ₂ O ₃	9.60	10.11	17.63	18.56	9.24	10.02	8.00	8.83	6.18	6.81		
FeO	0.79	0.42	0.17	0.18	0.14	0.15	0.71	0.78	0.24	0.26		
CaO	0.75	0.82	1.58	1.66	0.77	0.84	2.29	2.53	0.72	0.79		
MgO	2.64	2.10	3.16	3.32	3.26	3.54	3.30	3.64	3.69	4.05		
MnO	0.13	0.09	0.27	0.28	0.10	0.11	0.11	0.12	0.14	0.15		
Na ₂ O	0.82	0.45	2.59	2.62	3.43	3.73	3.33	3.68	3.72	4.09		
K ₂ O	3.24	0.96	1.87	1.90	1.77	1.92	1.85	2.04	2.41	2.66		
H ₂ O+	6.25	8.10	7.75	8.15	1.91	2.08	4.69	5.18	7.18	7.80		
H2O-	5.26	9.19	9.79	10.29	8.69	9.45	7.51	8.28	8.03	8.84		
CO ₂	-	0.42	0.09	0.09			0.24	0.26	0.34	0.37		
С	0.08	0.13			0.19	0.21	0.17	0.19				
P2O5	0.21	0.80	0.18	0.19	0.05	0.05	0.20	0.22	0.07	0.08		
SFe2S					6.52	7.08	3.59	3.96	2.71	2.98		
Σ	99.80	99.86	99.61	99.99	99,51	99.97	99.52	100.03	99.66	99.91		
SiO ₂ gw	10.20	-	2.20		5.94		7.34		7.42			
SiO ₂ am	1.75	-	2.21		1.49		1.50		1.38			

TABLE 2 Chemical Composition of Samples, Site 337

Note: Sample 2072 = (2-6, 100-102 cm), silty clay; Pleistocene; Sample 2077 = (4-2, 30-32 cm), tuff; Pliocene. Sample 344 = (6-2, 67-69 cm), volcaniclastic tuff; Sample 346 = (7-2, 75-77 cm), clay, pelagic, ferrimontmorillonitic; Sample 352 = (8-2, 76-78 cm), clay, pelagic diatoms, ferrimontmorillonite; Sample 370 = (12-3, 76-78 cm), clay, pelagic with zeolites.

The series is divided into three subseries based on the mineralogic composition of the sediments.

Subseries 21 (Cores 5-4 to 7-6, 42.2-66.0 m)

The upper boundary of the subseries is definite with overlying calcareous silts with foraminifera. The lower boundary is less distinct, overlying clays with an abundant siliceous microfossil content. The clays of Subseries 21 are yellow and often orange and brown. The upper part of the subseries contains volcaniclastic layers. The layers consist of aggregates, sometimes ferruginous glassy fragments, some altered to clay matter, and intergrown druses of phillipsite microcrystals. The phillipsite shows weak reflections 7.1, 5.3, 3.18 on the X-ray pattern (Sample 344). In the other samples, the clay contains sand-size fragments of basic plagioclase, aggregate quartz grains, fragments of brown and light volcanic glasses, and small prismatic apatite crystals (Samples 346, 348). In the lower part of the subseries, there is a layer of volcanic glass (Sample 350), consisting of a slightly crystallized mass with a peculiar curvilinear "pale-like" structure. Fragments of hornblende crystals are present within the slightly crystallized glass.

Clay Mineralogy

The characteristics of the ferrimontmorillonites of this subseries are specific. They are marked by very high values of basal reflections at $d(_{001}) = 15.5 - 15.6$ Å in

untreated samples, and high peaks with d = 18Å in saturated samples. This is a sign of a good mineral crystallization. The minerals belong to the Ca form. The heightened values of the basal reflection (d = 18Å) in glycerine-saturated samples may be attributed either to the overall finely dispersed nature of the mineral, or to the presence of mixed-layer clays. Apart from reflections with d(002) = 8.9Å-9Å belonging to the secondorder basal reflection of true montmorillonite, diffraction patterns of saturated samples show series of peaks in the 9.1-10.0Å interval. These reflections often merge to form a characteristic horizontal plateau (Sample 346). This suggests a continuous series of mixed-layer montmorillonite-hydromica units with a micaceous-toexpandable layer ratio varying from 0 to 10%. The content of the mixed-layer phase does not exceed 10%-15%. The usual weak resistance to acid treatment observed is also characteristic of the ferrimontmorillonite (Sample 348).

Along with montmorillonite and the series of mixedlayer clays genetically associated with it, there is also a true micaceous mineral present (Samples 347, 348). They are present in a limited quantity, not greater than 10%-15%. There is also a small admixture of defective chlorite with reflection d = 13.6-13.8Å. These are sometimes combined with 7.1Å on diffraction patterns of preheated samples, and a 7Å mineral preserved after boiling the samples in 10% HCl (kaolinite) (Sample 348).

Subseries 22 (Cores 8-1 to 11-5, 66.0-104.0 m)

The subseries consists of light-yellow, gray and brown-gray carbonate-free clays, with abundant remains of siliceous microfossils (Radiolaria, diatoms, and sponge spicules). These are present in the clay and increase with depth to the base of the subseries. The clay is a fine-aggregate and often nearly isotropic. Along with siliceous skeletons, the sediment contains numerous fragments of various volcanic glasses. Brown, diverse volcanic glass fragments (N = 1.605-1.610) prevail. They are characterized by angular contours, with a vesicular surface and rugged edges. In samples where they are especially abundant, there are also large fragments of fresh basic plagioclase crystals (Sample 354).

Clay Mineralogy

The clay retains a montmorillonitic composition, but its character is substantially altered. In the upper part of the subseries, where siliceous organisms are abundant, montmorillonite is represented by a wellcrystallized Ca-form with a mixed-layer unit (Samples 352, 354, 356). However, lower in the section it is replaced by a poorly crystallized form. The latter has abnormally high values of d(001), and wide basal peak profiles equal (in the saturated state) to 18.8-19.2Å. There also is a typical broad halo in the 3.0-4.5Å region, associated with an abundance of amorphous material (uncrystallized glass or opal, Samples 160, 162, 164). Abnormally high values of basal reflections and their diffuse shape are very probably due to the sorption, by montmorillonite, of amorphous SiO2. Samples boiled in soda indicate a distinctly decreased spacing of $d(\infty)$, and an improved overall configuration. However, even in this "regenerated" form, the montmorillonite of this subseries differs noticeably from its counterpart in the upper part of the subseries. It retains its poor crystallization and is devoid of any mixed-layer units. There is a further tendency for it to lose its iron content and to gain in SiO₂ (Table 2, Sample 252). Accessory clay minerals include: illite-type hydromica, defective chlorite, and kaolinite. This association is found with pelagic, essentially siliceous sedimentation, where, with authigenic forms of disorderly montmorillonite, terrigenous admixtures of other clay minerals are also deposited.

Subseries 33 (Cores 12-1 to 12-5, 104-113.0 m)

The subseries is thin and consists of light, grayishbrown clays, frequently deformed, and often transformed into a breccia together with small fragments of the altered basement basalt. The basic sediment nature is often nonuniform. Remnants of siliceous organisms decrease noticeably. Only large Radiolaria chambers are sporadically present. The sediment includes abundant fragments of crystallized glasses, quartz and feldspar grains, shreds of montmorillonite-chlorite aggregates, sometimes with very fine zeolite (?) microcrystals derived from altered underlying basalts. In the diffraction patterns of the clay and clay fractions, reflections with d = 2.97-2.98Å are constantly recorded. These belong to a Ca-Mg carbonate, also derived from the altered basalt basement (Samples 369, 371). The chemical analysis of clay from the least altered parts of the core indicates a decrease in iron content and an increased silica content (Table 2, Sample 270). However, the clay as a whole is characterized by a sharp predominance of well-crystallized montmorillonite of the Ca-form in combination with a series of mixed-layer units genetically related to it. In the mixed-layer clays, interlayering units dominate in which a micaceous phase is present in amounts of 10%-20% to 90%. This is expressed in the diffraction patterns of saturated samples by characteristic sharp peaks with d = 9.1-9.3Å and d = 9.8-9.9Å (Samples 368, 369, 370). Sometimes weak reflections with d = 8.9, 7.6, 3.19Å are present, suggesting zeolites of the clinoptilolite series.

On the whole, the mineral association of the subseries shows noticeable traces of interaction of the sediments with the underlying basement of hydrothermally altered basalt.

PRINCIPAL RESULTS AND CONCLUSIONS

1. Two sediment series corresponding to DSDP unit subdivisions have been identified and investigated. A more detailed subdivision of the series into subseries based on mineralogic data has been discussed.

2. Although Site 337 is located on the top of rift mountains at the eastern edge of "extinct axis" in the Norwegian Basin, the location did not have a significant effect on the sedimentation of the Pleistocene deposits. The sediments were derived via ice-rafting and erosion of the floor of the Norwegian Basin.

An intensive supply of "exotic" sedimentary material by ice-rafting is apparent by the abundance in Pleistocene deposits of unsorted sand-silt and clays, with pebbles of granites, siltstones, limestones, biotite, and amphibolite schists. A higher sedimentation rate is indicated in the "Glacial" as compared with the Tertiary.

This sedimentation, associated with "glaciation" has affected the formation of the clay minerals in the Pleistocene section. A major role was played by the sediment components supplied by ice-rafting, specifically the abundance of micas which made up to 70% of the clay fraction. They are represented by dioctahedral illite-type hydromica (predominantly without expandable layers or rarely with more than 10%-20% expandable layers). The polytype modification of hydromicas is mostly 1 Md, and partly 1 M.

Biotite schists with unstable trioctahedral micas were an especially important component. These micas were transformed to a series of alteration products: mixedlayer vermiculite-montmorillonite minerals, defective chlorites, and K-montmorillonites. Petrographic observation also shows all stages of corrosion and alteration of the biotite plates in situ, and fully corroborates Xray data. The presence of this clay minerals assemblage with hydromica, chlorite, an admixture of montmorillonite, kaolinite, and mixed-layer units (with the same ratio of components) was previously noted in the redeposited glacial sediments in the Barents and White seas (Rateev, 1949; Kalinenko and Rateev, 1974). This is the basis for the conclusions that in the Pleistocene clay minerals were ice-transported to the Norwegian-Greenland Sea, either from the north (Spitsbergen) or

from the east (Scandinavia). A study of the clay minerals distribution by sediment types indicated that the pelagic sediments of the Oligocene/Pliocene have a montmorillonite composition for the clay fraction, frequently with zeolite. However, in the Pleistocene sediments the clay fraction has hydromicas (60%), and chlorite, with admixture of vermiculitemontmorillonite mixed-layer units (30%).

Therefore, the ice transport of clastic and clay minerals and the abundance of biotite are the main characteristics of the Pleistocene sediments. Clastic clay material was supplied extensively during the interglacial periods. During these periods, the hydromica composition of the clay fraction of the Pleistocene sediments was drastically increased. However, during the periods of glaciation volcanogenic montmorillonite increased. This is not a universal, world-wide phenomenon. For example, in the Black Sea, Stoffers and Muller (1972) established an inverse relationship. In the Black Sea sediments of the "glacial" epoch, an increase of montmorillonite was noted.

3. Series 2 consists of typical pelagic clays, containing abundant remains of siliceous microfossils in the middle subseries. The clay minerals formed by alteration of hyaloclastics and pyroclastics of tholeiitic basalts, and represented by characteristic ferrimontmorillonite. It is characterized by a high iron content (Fe₂O₃ to 10.5%), an approximately equal proportion of K₂O and Na₂O, and a very low resistance to acid treatment. With increasing depth in the series section, a slight drop in iron content and an increase in silica and in the sum total of alkalis was observed (see Table 2). Despite a similar chemical composition and that all ferrimontmorillonites belong to the Ca-form marked by high values of basal reflection with $d(001) = 15-16\text{\AA}$, two mineral types are clearly discernible.

The first is characterized by a high degree of crystallization and a presence of genetically related peculiar series of mixed-layer hydromica-montmorillonite units with a "complete set" of mica-toexpandable layer ratios varying from 0 to 100%. The other mineral which is present only in the middle, siliceous subseries is free of any mixed-layer units and is poorly crystallized. Characteristics of the basal reflection (up to 19.8Å in saturated samples) indicates a "littering" of the montmorillonite interlayers with amorphous silica.

4. Volcaniclastic tuffs and clays with phillipsite are present at the boundary between the subseries. Femontmorillonite has a high Fe_2O_3 content (18.55%) (Table 2). It was formed as a result of inflow of ironcontaining solution enriched in manganese. The manganese is present in the clay, and as isolated manganese concretions at the interface between series.

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