

6. MINERALOGY AND PETROGRAPHY OF LEG 46 BASALTS

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INTRODUCTION

Leg 46 basalts were recovered at Hole 396B, approximately 160 km east of the Mid-Atlantic Ridge. Basalts 13 million years old, overlain by 150 meters of sediment, have been divided on board ship into eight lithological units, subsequent to petrographic, magnetic, and chemical studies. The upper lithological unit (165 m) is composed of pillows, except for Unit 3, which is a large cooling unit. The lower part is built up of pillow breccias and hyaloclastites. The freshest basalts, slightly vesicular, more often with plagioclase and olivine phenocrysts, were sampled from Units 1 to 4 for chemical and mineralogical studies: 32 basalts were examined in thin section and their bulk chemistry was determined. Electron microprobe analyses were performed on pyroxene, plagioclase, olivine, and titanomagnetite from 12 representative samples.

PETROGRAPHY

Texture, secondary products, and the size variation of the magmatic minerals are shown in Table 1. Thin sections of the basaltic rocks show varying degrees of alteration. The groundmass is always transformed, in varying degree, into smectites. Vesicles are filled with smectites and locally with calcite. Some cross-cutting veins show a similar mineralogy, although zeolites may occur. The two more abundant magmatic minerals, plagioclase and clinopyroxene, are always well preserved. Calcite occurs in cracks in some plagioclase phenocrysts. Olivine is locally transformed into smectites and/or calcite.

Units 1 and 2, separated by interbedded carbonate sediments, consist of fine-grained basalts with sparse olivine and plagioclase microphenocrysts which represent less than 1 per cent of the rocks. Glomerocrysts of these minerals sometimes occur. Plagioclase, olivine, and clinopyroxene are scattered in an abundant groundmass (nearly 30%). In the pillow, rim textures vary from glassy through spherulitic to variolitic, as in Sample 7-2, 42-45 cm. Varioles, small (0.2 mm diameter) in the outer zone, increase in size until (at 0.4 mm diameter) they become coalescent. In the pillow core, hyalopilitic and hyalo-ophitic textures are the more developed.

Unit 3 is a cooling unit. Macroscopically it is composed of dark gray, medium-grained phryic basalts with sparse olivine and plagioclase phenocrysts; texture is intersertal. The two borders are fine grained.

Lithologic Unit 4, more altered than the units above, is built up of a porphyritic pillow sequence. These basalts are gray and fine grained with 15 to 20 per cent phenocrysts.

Plagioclase phenocrysts are five times more abundant than olivine phenocrysts. Plagioclase and olivine glomeroporphyritic clots are locally present. Brown chrome spinel occurs as inclusions in plagioclase and as isolated phenocrysts in the groundmass. Plagioclase microlites with pronounced skeletal shape are locally oriented in the abundant mesostasis.

We have examined one sample from lowermost Unit 5. It is grayish brown fine-grained basalt with many patches and veins of calcite and zeolites. The texture is hyalopilitic and trachytic.

CHEMISTRY

Analytical Method

Major elements of 32 basalts have been determined by P. Cambon at the COB using an X-ray fluorescence method similar to that used for shipboard analyses (Bougault and Cambon, 1973; Bougault, in press). Glass and varioles (39 analyses) from Sample 7-2, 73-75 cm, were analyzed using the microprobe. The analytical method is similar to that employed for mineralogical determination (see below). The results are given in Table 2, where chemical and lithologic units are also specified. The bulk chemistry is characteristic of abyssal tholeiites. MgO and SiO₂ are nearly constant, and K₂O is always low in every unit.

Units 2 and 3 (A₃) contain more Ti, Fe, P, and Mn than Unit 1 (A₁-A₂). In Unit 1, the upper part (A₁) is poorer in Ti than the lower A₂. The basalts of Unit 4 (B) are high-alumina tholeiites if the 16.4 per cent Al₂O₃ limit is used (Miyashiro et al., 1969). The Al₂O₃ and CaO enrichment, coupled with low FeO, TiO₂, P₂O₅, and MnO contents when compared with Unit A, may be correlated at first with the abundance of the plagioclase phenocrysts. The fundamental differences in Al₂O₃ and TiO₂ contents between Units A and B and their sub-units may be easily checked in the Al₂O₃-TiO₂ diagram (Figure 1).

The correlation between vertical position and differentiation is inverse in Units A and B. Unit A shows an apparent TiO₂ increase coupled with a slight Al₂O₃ decrease towards the base of the unit: a differentiation inverse to the eruption sequence. In contrast, a normal differentiation pattern occurs in Unit B, in which TiO₂ increases and Al₂O₃ decreases towards the top of the volcanic pile. In each A sub-unit the differentiation trend is similar: TiO₂ increases toward the top of each A sub-unit, especially in A₂, thus indicating a normal differentiation trend with depth. The apparent reversal evolution of Unit A may be a consequence of fractional crystallization of slightly different magma batches. In Unit B, by contrast, as Al₂O₃ decreases and TiO₂

TABLE 1
Summary of Petrographic Features of Basalts From Hole 396B

Lithologic Unit Samples (Interval in cm)	Texture	Phenocrysts	Plagioclases		Olivine		Glassy Groundmass			Alteration Products	Vesicle Diameter
			Microphenocrysts	Microlites	Microphenocrysts	Microlites	Clinopyroxenes	Opaques	More or Less Altered		
Unit 1											
5-2, 134-136, #15	Intersertal to hyalo-ophitic		Glomerocrysts 0.8/0.4 mm	0.2/0.03 mm A	Some euhedral crystals, sometimes rounded, 0.6 mm A	0.04 mm	0.05 mm A	Granules	+	Calcite smectite	+
7-1, 106-108, #10	Halo-ophitic			0.2/0.05 mm A		0.05 mm anhedral A	Cerviconic and axiolitic A	Granules	+	Calcite	
7-2, 42-45, #5	Glassy-variolitic to spherulitic; varioles size 0.2 to 0.4 mm			0.2/0.3 mm A	0.8/0.2 mm A	+	+	+	+ Varioles A Glass A	Calcite	+
7-2, 73-75, #7B	Intersertal			0.6/0.02 mm	1.6/0.7 mm rounded		+	+	+	Smectite	
8-2, 43-45, #4B	Hyalo-ophitic			0.6/0.4 mm A		0.1 mm A	Axiolitic A	Granules	+	Smectite	+
9-1, 40-49, #60	Hyalo-ophitic		0.6/0.2 mm	0.3/0.02 mm A		0.1 mm A	Dendritic A	Titanomagnetite A	+		0.1 mm
9-3, 14-16, #2A	Intersertal		0.8/0.2 mm A			0.2 mm A	Axiolitic A 0.2/0.01 mm	0.2 mm	+	Smectite calcite	+
10-1, 43-46, #8	Very sparsely phryic, hyalo-ophitic		2.1/0.2 mm			0.03 mm	Axiolitic	Granules	+	Smectite	
10-2, 59-61, #9	Very sparsely phryic, hyalo-ophitic	6/3 mm with melt inclusions		0.7/0.02 mm		0.1 mm	Axiolitic	+	+	Smectite calcite	
11-1, 87-99, #108	Intersertal			0.5/0.02 mm		0.05 mm skeletal	Axiolitic	Granules	+	Smectite calcite	
11-2, 18-20, #2	Very sparsely phryic, intersertal			0.5/0.03 mm A	0.5 mm	0.5 mm	Axiolitic A	Granules	+	Smectite	0.2 mm
12-1, 110-112, #7	Very sparsely phryic, intersertal		0.9/0.6 mm	0.6/0.04 mm		0.2 mm	Axiolitic	Granules	+	Smectite	
Unit 2											
13-2, 137-140, #13	Spherulitic		0.2/0.02 mm bow-ties			0.04 mm skeletal	Dendritic	Granules	+	Smectite calcite in veins	0.2 mm
14-1, 16-21, #3	Very sparsely phryic, spherulitic		0.5/0.03 mm bow-ties		1.2/1.05 mm	0.2/0.04 mm	Axiolitic	Granules	+	Smectite	0.2 mm vesicles and mafic spherules
14-2, 113-115, #11	Intersertal		4.3/0.1 mm A			0.15 mm granular	Axiolitic, fan-shaped; 0.1 mm A	Grains titanomagnetite A	+	Smectite	0.2 mm
14-3, 70-72, #5C	Very sparsely phryic, intersertal		0.9/1.07 mm	0.7/0.05 mm		0.05 mm	Axiolitic	Grains euhedral	+	Smectite	
Unit 3											
15-1, 120-128, #12E	Intersertal			1/0.05 mm			Axiolitic, fan-shaped	Granules	+	Smectite	
15-2, 79-81, #17	Intergranular		1.6/1 mm	2/0.1 mm		0.5 mm granular A	0.5 mm granular A	Grains 0.03 mm A	+ A	Smectite	0.2 mm
15-2, 99-107, #2A	Intersertal to intergranular		1.2/0.5 mm	1.8/0.1 mm		0.1 mm grains	0.14 mm grains	Grains skeletal 0.14 mm	+	Smectite calcite	0.4 mm
15-4, 114-116, #7	Intersertal		2.5/1.5 mm A	0.8/0.05 mm A		0.14 mm grains skeletal A	0.25 mm grains A	Grains 0.07 mm skeletal	+	Smectite	0.3 mm
15-5, 16-18, #2B	Intersertal		3/1.4 mm	0.2/0.08 mm		Grains 0.07 mm	Grains 0.14 mm skeletal	Grains 0.07 mm	+	Smectite calcite	0.2 mm
Unit 4											
16-1, 28-30, #4	Porphyritic; hyalo-ophitic to intersertal	3/1 mm euhedral melt inclusions	0.5/0.03 mm	2 mm rounded A		0.14 mm	Axiolitic A	Oxides granules Cr-spinel	+	Calcite smectite	0.4 mm
16-3, 34-41, #4B	Porphyritic; hyalo-ophitic	5/3 mm melt inclusions	0.4/0.03 mm	2 mm euhedral or rounded		0.15 mm	Axiolitic 0.07 mm	Granules	+	Smectite calcite zeolites	0.4 mm
18-1, 55-57, #4C	Porphyritic, intersertal	3/1 mm melt inclusions	0.6/0.04 mm A	1 mm rounded		0.1 mm A	Axiolitic A	Oxides granules and Cr-spinel 0.07 mm	+	Smectite calcite	0.14 vesicles and spherules 1 mm
20-1, 65-69, #4	Porphyritic; hyalo-ophitic	5/3 mm	0.2/0.02 mm	3 mm rounded		0.16 mm	Axiolitic	Grains 0.04 mm	+	Smectite calcite	0.6 mm vesicles and mafic spherules; 0.4 mm

20-5, 84-93, #9C	Porphyritic, hyalo-ophitic	5/2 mm melt inclusions	0.4/0.03 mm	1.5 mm	Oxides granules 0.14 mm and Cr-spinel	Axiolitic 0.17 mm rounded	Calcite smectite zeolites in veins 0.4 mm
21-1, 80-91, #10	Porphyritic, hyalo-ophitic to interstitial	3/1.5 mm glomerophenocrysts	0.4/0.03 mm	1 mm	Oxides in grains and skeletal Cr-spinel	Axiolitic 0.02 mm rounded	Smectite calcite 0.1 mm
21-2, 16-24, #2	Porphyritic, hyalo-ophitic	3/1.5 mm melt inclusions	0.04/0.03 mm	1 mm rounded	Oxide grains and Cr-spinel	Axiolitic A 0.1 mm	Smectite calcite 0.1 mm
22-2, 61-73, #3C	Porphyritic, hyalo-ophitic	2.5/1 mm glomerophenocrysts A	0.4/0.03 mm	1.5/1 mm calccalcic A	Oxides grains and Cr-spinel	Axiolitic A 0.2 mm	Smectite calcite 0.4 mm and mafic spherules
22-2, 110-120, #7C	Porphyritic, hyalo-ophitic	3/1 mm melt inclusions A	0.4/0.03 mm	1 mm rounded A	Oxides and Cr-spinel	Axiolitic, fan-shaped A 0.4 mm A	Calcite smectite
22-3, 104-107, #7B	Porphyritic, hyalo-ophitic	3/1.5 mm	0.5/0.03 mm	1 mm	Granules	Axiolitic 0.2 mm rounded	Calcite smectite zeolites in veins 0.2 mm
Unit 5		Pilotaxitic fluidal	0.4/0.02 mm how-text and microfolds		Dendritic 0.4 mm skeletal	Granules	Calcite zoilites smectite in veins
24-1, 51-53, #9	Pilotaxitic fluidal						

increases from B_2 to B_1 — towards the top — B_1 may have been derived from B_2 through a plagioclase fractionation.

Variation diagrams relating FeO/MgO to Fe and Ti oxides show features similar to those noted above (Figure 2). FeO/MgO (1 to 1.5) increases towards the bottom in Unit A and normally towards the top in Unit B. Glass and varioles from Sample 396B-7-2, 42-45 cm are also shown in Figure 2 (by the areas enclosed by solid lines) and reported in Table 3; the details of these areas appear in Figure 3. The FeO/MgO ratios of varioles and glass are lower than that of the host rock but similar to those of the Unit A basalts. MgO depletion in the slightly altered pillow rim would explain in the low FeO/MgO ratio of the host rock (Honnerez and Bohlke, in press). Varioles are richer in TiO_2 than glass and host rock. Fe distribution is similar in varioles and glass; Fe oxide increases with ratio increase. This trend shown by varioles and glass in a single pillow is similar to the normal evolution of abyssal tholeiites. The narrow compositional range between varioles and basalts excludes an immiscibility process for the variole genesis. Undercooling is the more probable explanation of their formation (Lofgren, 1974).

PHASE CHEMISTRY

Electron microprobe analyses were obtained for 12 samples of lithologic Units 1 to 4. The minerals were analyzed using the CAMECA MS 46 microprobe at the University of Paris. The analytical conditions were as follows: accelerating voltage 15 kV, specimen current 30 nA, counting time 50 seconds for olivine and pyroxene and 25 seconds for plagioclase. Natural minerals were used as standards. The microprobe data were reduced on the CIRCE IBM 370 computer employing the EMPEDAR program (Rucklidge and Gasparini, 1969). In Sample 396B-7-2, 42-45 cm, the fine grain size of the pyroxenes does not permit their analyses. This is also true for opaque oxides in many samples.

We analyzed 243 pyroxene samples, using the microprobe. Representative selected samples are reported in Table 4. The pyroxenes, slightly brown in thin section and scattered in the groundmass, constitute one of the last crystallizing phases. The habit is axiolitic, dendritic, and comb-like (except in the cooling unit of Core 15, where it is more often granular). These pyroxenes fall always in the augite-salite field for Units 1, 2, and 4. The pyroxenes of the cooling unit are restricted to the augite field. Al_2O_3 and TiO_2 contents are high and show wide variations throughout the unit. TiO_2 contents of Units 1 and 2 are the highest. In the porphyritic basalt of Unit 4, Al_2O_3 content reaches the highest values (up to 8.69%).

In the Ca-Mg-Fe diagram (Figure 4, pyroxenes of Unit 1 fall in the same field, despite the dispersion. There is no obvious trend, only a slight Fe-Mg variation. Sample 396B-5-2, 134-136 cm exhibits a Mg-Ca variation. In Unit 2, pyroxenes of the analyzed sample fall in the same area, but Mg is nearly constant. In Unit 3, pyroxenes fall in the augite field and are relatively more magnesium-rich than pyroxenes of the unit above. An iron-enrichment trend at nearly constant Mg is evident for the two analyzed samples. Such a trend is usually considered a result of rapid metastable crystallization linked to undercooling (Kirkpatrick, 1976). This trend is curiously well-developed in this cooling unit

TABLE 2
Chemical Analyses of the Basalts From Hole 396B

Sample	5-2 134-136	7-1 #10	7-2 #5	7-2 #7B	8-2 #4B	9-1 #6D	9-3 #2A	10-1 #8	10-2 #9	11-1 #10B	11-2 #2	12-1 #7	13-2 #13	14-1 #3	14-2 #11
Lithological Unit	1	1	1	1	1	1	1	1	1	1	1	1	1	2	2
Chemical Unit	A ₁	A ₂	A ₃	A ₃											
SiO ₂	49.46	49.90	48.82	49.41	49.44	48.99	50.22	49.89	49.22	49.19	49.46	50.01	49.23	48.91	49.52
Al ₂ O ₃	15.22	15.72	16.02	15.24	15.15	15.95	15.50	14.91	15.23	15.06	15.03	14.93	15.22	15.21	15.09
Fe ₂ O ₃ ⁺	10.37	10.01	10.93	10.14	10.45	11.24	9.92	10.31	10.53	10.74	10.55	10.58	10.72	11.28	11.11
MnO	0.17	0.17	0.18	0.16	0.17	0.18	0.15	0.16	0.17	0.17	0.17	0.16	0.18	0.18	0.17
MgO	8.08	7.77	6.80	8.42	8.03	6.89	7.10	7.60	7.38	7.75	7.50	7.65	7.74	7.80	7.37
CaO	12.00	12.24	12.29	11.80	11.57	12.26	11.93	11.51	11.88	11.59	11.55	11.48	11.65	11.23	11.11
Na ₂ O															
K ₂ O	0.28	0.24	0.17	0.23	0.27	0.27	0.23	0.28	0.26	0.24	0.24	0.26	0.15	0.26	0.26
TiO ₂	1.42	1.43	1.48	1.41	1.52	1.61	1.59	1.56	1.53	1.54	1.54	1.52	1.55	1.67	1.65
P ₂ O ₅	0.15	0.13	0.16	0.13	0.14	0.18	0.15	0.15	0.15	0.14	0.15	0.13	0.15	0.18	0.17
L.O.I. 110°	0.70	0.29	0.87	0.65	1.42	0.93	0.50	1.36	0.59	0.64	0.41	1.21	0.40	0.66	0.35
L.O.I. 1050°	0.38	0.59	0.66	0.46	1.24	0.70	0.64	0.88	0.67	0.38	0.45	1.01	0.38	0.65	0.68

composed of the largest minerals. Nevertheless, the presence of altered glass testifies to the efficiency of rapid cooling. This trend is also recognizable in Unit 4 pyroxenes, which are slightly richer in Ca and Mg.

Plagioclases

Selected plagioclase analyses, reported in Table 5, have no K-feldspar component. Plagioclase microlites show a narrow compositional range of An₆₃₋₇₃. The highest An values occur in Unit 2. In Unit 3 two generations of feldspar may be present. The big ones are calcic (An₆₅₋₇₁) than the small ones (An₄₇₋₅₁). The plagioclase phenocryst of Unit 4 has a nearly constant composition in the core (An_{75-85; 75-89}); the narrow rim is more sodic (An_{64-68; 70-75}) and shows values similar to those of the coexisting microlites (An₆₅₋₇₃).

Olivines

Selected olivines are reported in Table 6. Olivine is magnesian in all the units (Fo₇₈₋₈₈). Unit 4 exhibits the more magnesian olivine (Fo₈₃₋₈₈). Olivine phenocrysts do not appear to be significantly zoned, and have the same composition as the olivine microlites.

Opaque Oxides

Titanomagnetites were analyzed in Samples 396B-9-1, 40-49 cm and 396B-14-2, 113-115 cm. Partial analyses are summarized in Table 7. The FeO-TiO₂ contents are similar to those reported by Mazzullo et al. (1976).

CONCLUSION

1. Bulk chemistry and chemical data are consistent with the lithologic and chemical unit defined aboard ship. Basalts are typically abyssal tholeiites; the Unit 4 porphyritic basalts are high-alumina tholeiites.

2. Unit A is more fractionated than Unit B. Unit A shows an apparent inverse evolution with depth, whereas in Unit B the upper basalt may have evolved from the lower basalt through a plagioclase fractionation.

3. Varioles have a compositional range similar to that of the basalt, and were formed by an undercooling process.

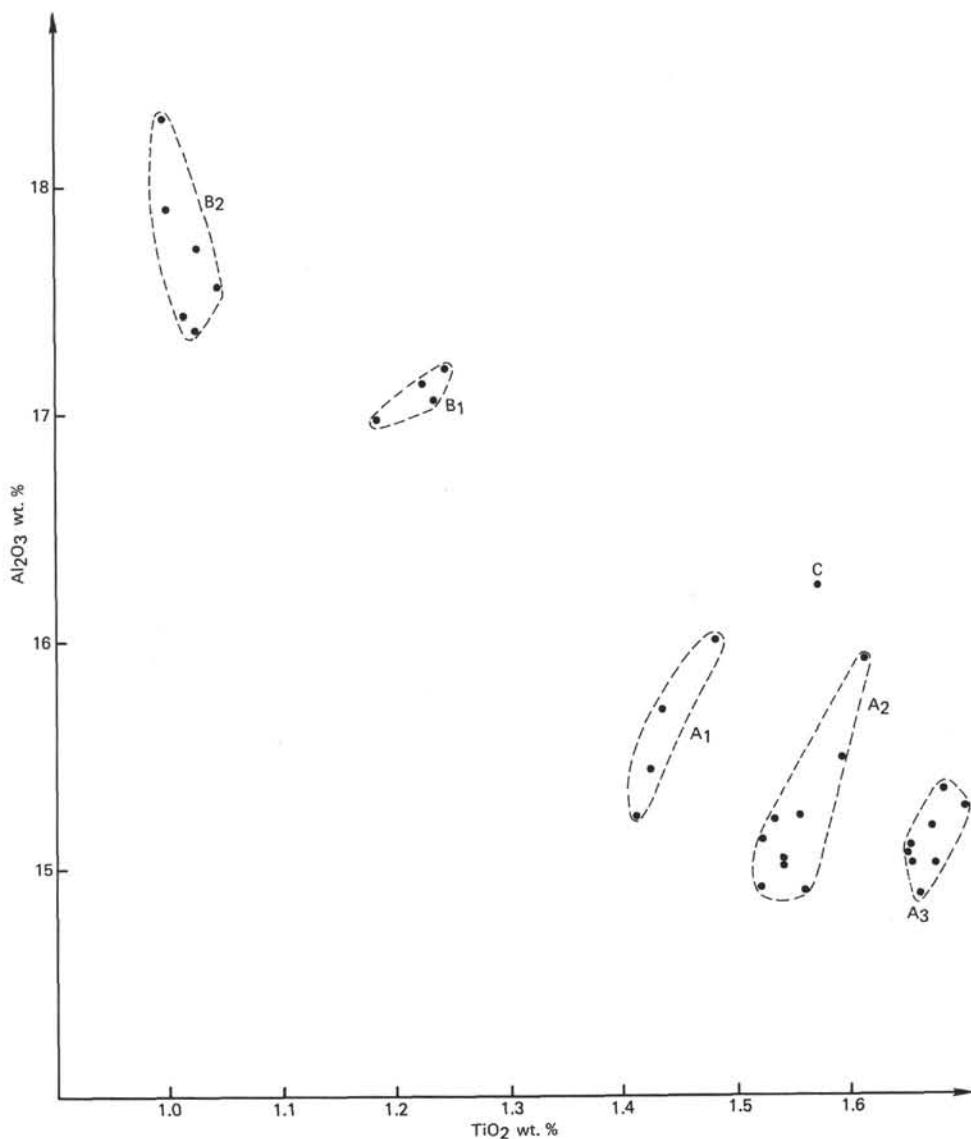
4. Augite and salite from Units 3 and 4 show a Fe-Ca variation at nearly constant Mg, a feature characteristic of metastable crystallization which occurs during undercooling.

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TABLE 2 - *Continued*

14-3 #5C	15-1 #12E	15-2 #1J	15-2 #2A	15-4 #7	15-5 #2B	16-1 #4	16-3 #4B	18-1 #4C	20-1 #4	20-5 #9C	21-1 #10	21-2 #2	22-2 #3C	22-1 #7C	22-3 #7B	24-1 #9
2	4	4	4	4	4	4	4	4	4	4	4	4	4	4	4	5
A ₃	B ₁	B ₁	B ₁	B ₁	B ₂	C										
50.19	48.86	49.25	49.71	49.78	49.59	49.47	48.97	49.29	48.82	48.62	48.78	49.14	48.36	49.17	49.42	49.10
15.30	15.37	14.91	15.05	15.12	15.03	17.21	17.15	17.07	16.99	17.74	17.57	17.44	18.33	17.91	17.37	16.28
10.68	11.25	11.04	11.05	10.86	10.93	9.19	9.46	9.35	9.25	8.57	8.72	8.65	8.80	8.54	8.61	10.59
0.16	0.19	0.17	0.16	0.17	0.18	0.15	0.15	0.15	0.15	0.14	0.14	0.13	0.15	0.14	0.14	0.17
7.04	7.10	7.96	7.78	7.87	7.28	6.22	7.12	6.90	7.98	7.89	7.42	7.49	7.89	7.20	7.93	7.29
11.33	11.68	11.08	11.10	11.08	11.19	12.72	12.48	12.53	12.44	12.62	12.74	12.54	12.14	12.44	12.51	11.83
0.27	0.28	0.09	0.17	0.18	0.32	0.17	0.20	0.21	0.22	0.14	0.20	0.17	0.28	0.10	0.12	0.26
1.70	1.68	1.66	1.65	1.65	1.67	1.24	1.22	1.23	1.18	1.02	1.04	1.01	0.99	0.99	1.02	1.57
0.17	0.17	0.16	0.16	0.16	0.16	0.13	0.13	0.12	0.12	0.10	0.10	0.11	0.12	0.11	0.10	0.18
1.63	1.00	0.33	0.90	0.87	1.20	1.21	1.54	0.99	0.77	0.90	0.70	0.78	1.25	0.72	0.63	0.66
1.17	1.47	0.04	0.99	0.60	0.82	1.15	1.20	0.88	0.50	1.09	1.12	1.11	2.75	1.21	0.80	0.56

Figure 1. *Al₂O₃-TiO₂ diagram.*

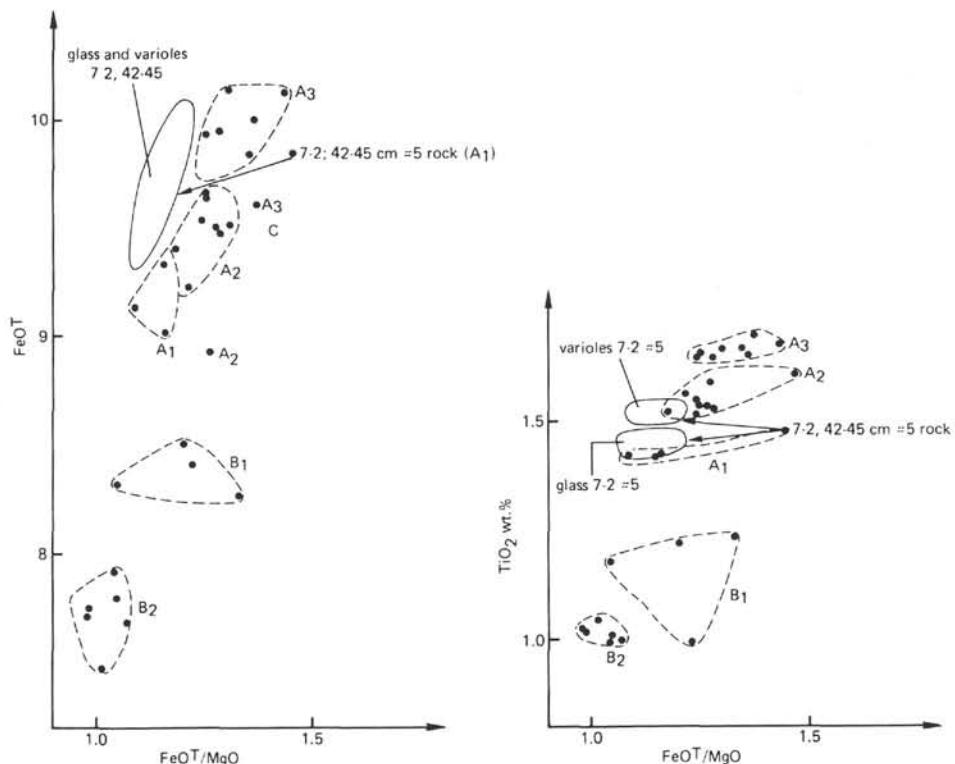
Figure 2. $FeOT$ and TiO_2 versus $FeOT/MgO$.

TABLE 3
Chemical Composition of Glass and Varioles of
Basalt Sample 396B-7-2, 42.45 cm; Analyzed by Microprobe

Sample	Glasses			Varioles			
				Rim	Core		
SiO_2	49.99	49.70	50.59	49.08	49.91	49.41	48.68
Al_2O_3	16.72	17.18	16.01	16.82	17.58	16.20	16.10
$Fe_2O_3^+$	9.55	9.48	9.65	10.22	10.98	10.12	9.72
MnO	0.23	0.22	0.23	0.26	0.26	0.52	0.23
MgO	8.84	8.76	8.72	9.01	6.08	8.60	8.28
CaO	11.24	11.34	11.83	11.11	11.58	11.44	12.05
Na_2O	2.35	2.18	2.16	2.25	2.29	2.59	2.59
K_2O	0.15	0.13	0.10	0.09	0.17	0.14	0.08
TiO_2	1.44	1.48	1.52	1.34	1.57	1.64	1.54
Total	100.52	100.47	100.81	100.18	100.42	100.67	99.27

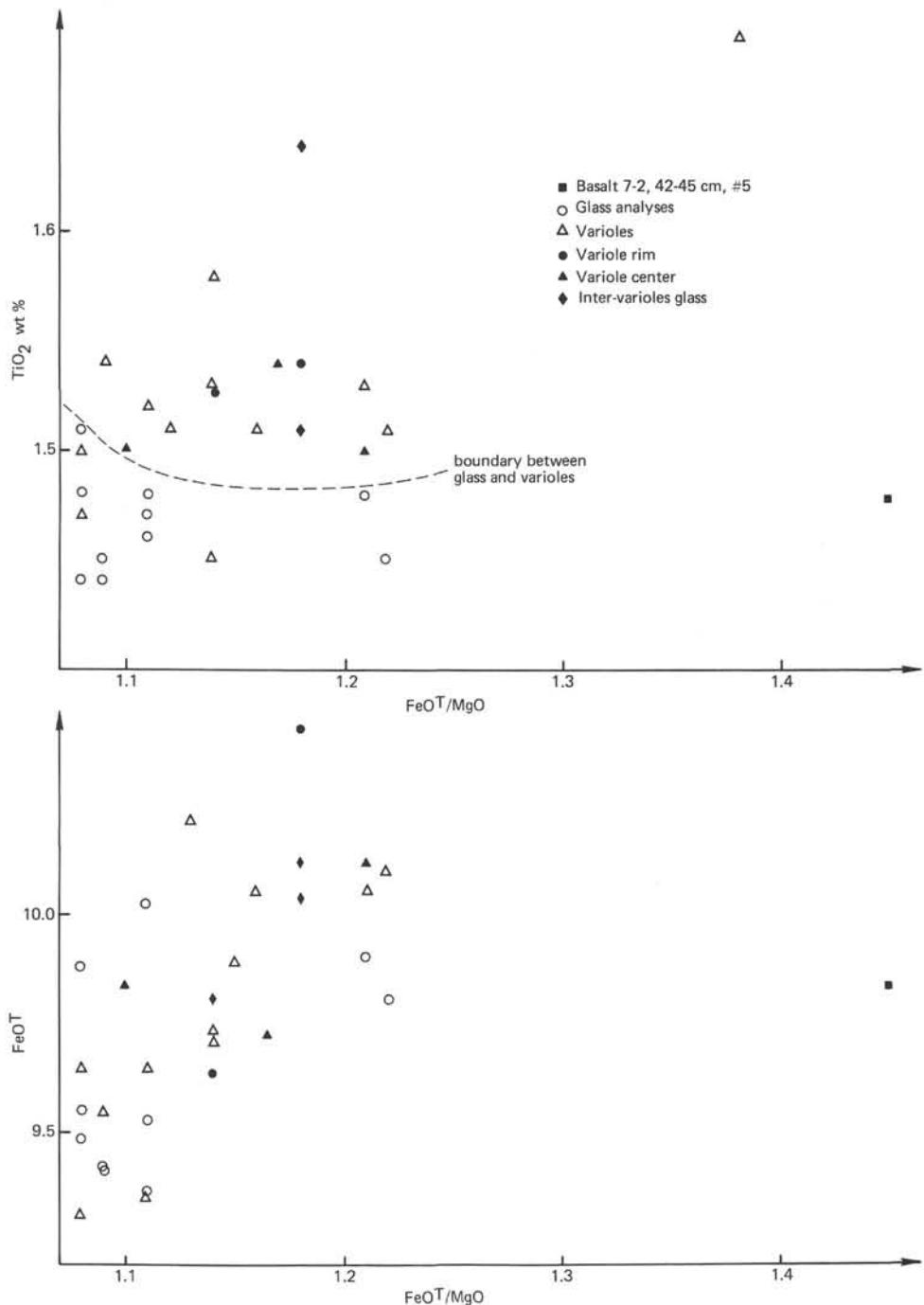


Figure 3. FeOT and TiO_2 versus FeOT/MgO for varioles and glass from Sample 7-2, 42-45 cm basalt.

TABLE 4
Representative Pyroxenes Analyzed by Microprobe

Sample	5-2, 134-136 cm		7-1, 106-108 cm		8-2, 43-45 cm		11-2, 18-20 cm		14-2, 113-115 cm	
SiO ₂	47.87	48.00	48.48	47.01	47.94	44.48	48.41	44.93	45.40	47.22
Al ₂ O ₃	5.74	5.21	6.08	5.61	5.38	5.29	3.58	4.78	5.27	5.06
FeO ⁺	11.43	8.46	12.72	12.81	9.32	13.72	12.19	13.52	11.51	12.85
MnO	0.30	0.27	0.38	0.32	0.24	0.32	0.35	0.40	0.33	0.33
MgO	13.29	15.47	9.76	12.18	13.61	10.17	12.11	11.67	12.01	12.70
CaO	19.36	19.49	19.60	19.73	21.41	21.88	19.82	21.54	20.00	19.85
Na ₂ O	0.46	0.36	0.46	0.43	0.49	0.50	0.43	0.61	0.42	0.51
TiO ₂	2.12	1.70	2.53	1.96	1.66	2.55	2.14	2.26	4.07	2.30
Total	100.57	98.95	100.01	100.06	100.05	99.05	99.04	99.70	99.00	100.81
Si	1.796	1.807	1.835	1.790	1.802	1.742	1.855	1.746	1.748	1.787
Al ^{IV}	0.204	0.193	0.165	0.210	0.198	0.244	0.145	0.219	0.239	0.213
Al ^{VI}	0.005	0.038	0.106	0.042	0.041	—	0.017	—	—	0.013
Ti	0.060	0.048	0.072	0.056	0.047	0.078	0.062	0.066	0.118	0.065
Fe	0.359	0.266	0.403	0.408	0.293	0.450	0.391	0.439	0.370	0.407
Mn	0.010	0.009	0.012	0.010	0.008	0.012	0.011	0.013	0.011	0.011
Mg	0.743	0.868	0.551	0.691	0.763	0.594	0.692	0.676	0.689	0.716
Ca	0.778	0.786	0.795	0.805	0.862	0.918	0.814	0.897	0.825	0.805
Na	0.034	0.026	0.034	0.032	0.036	0.038	0.032	0.046	0.032	0.038
Wo	41.4	40.9	45.5	42.3	44.9	46.8	42.9	44.6	43.8	41.8
En	39.5	45.2	31.5	36.3	39.8	30.3	36.5	33.6	36.6	37.1
Fs	19.1	13.9	23	21.4	15.3	22.9	20.6	21.8	19.6	21.1
										19.5

TABLE 4 – *Continued*

15-2, 79-81 cm				15-4, 114-116 cm				16-1, 28-30 cm				18-1, 55-57 cm				22-2, 61-73 cm				22-2, 110-120 cm				
48.84	46.50	47.48	47.43	47.48	46.98	47.92	47.64	48.53	46.12	48.28	47.31	45.63	46.45	46.89	47.69									
2.92	6.09	3.79	5.63	3.50	3.70	4.18	5.59	3.25	5.95	3.25	5.55	6.79	7.01	6.89	3.10									
14.48	10.08	11.85	9.68	14.09	16.99	9.00	10.73	13.36	10.47	13.65	11.48	9.02	7.83	8.57	14.87									
0.47	0.26	0.36	0.30	0.42	0.54	0.31	0.58	0.38	0.27	0.46	0.32	0.24	0.22	0.22	0.35									
13.77	12.95	13.74	12.82	12.65	12.14	12.71	13.78	14.30	12.80	14.48	12.40	12.52	12.25	13.40	14.07									
17.82	21.29	19.57	20.94	18.50	16.46	21.88	19.62	18.38	21.10	18.08	20.50	22.87	23.28	21.33	18.03									
0.43	0.48	0.45	0.53	0.45	0.46	0.66	0.50	0.46	0.30	0.33	0.46	0.48	0.33	0.37	0.33									
1.66	2.48	1.95	2.17	1.92	2.19	2.37	2.25	1.44	2.34	1.71	2.08	1.58	2.20	2.07	1.21									
100.39	100.12	99.19	99.51	99.02	99.46	100.03	100.69	100.10	99.35	100.23	100.10	99.12	99.41	99.74	99.65									
.																								
1.856	1.758	1.818	1.796	1.835	1.823	1.804	1.786	1.844	1.760	1.834	1.793	1.744	1.751	1.762	1.834									
0.131	0.242	0.171	0.204	0.159	0.169	0.196	0.214	0.146	0.240	0.146	0.207	0.256	0.249	0.238	0.140									
–	0.029	–	0.047	–	–	0.034	0.033	–	0.027	–	0.041	0.050	0.064	0.067	–									
0.048	0.070	0.056	0.062	0.056	0.064	0.067	0.063	0.041	0.067	0.049	0.059	0.045	0.063	0.058	0.035									
0.460	0.319	0.379	0.307	0.455	0.551	0.283	0.336	0.424	0.334	0.434	0.364	0.288	0.248	0.269	0.478									
0.015	0.008	0.012	0.010	0.014	0.018	0.010	0.018	0.012	0.009	0.015	0.010	0.008	0.007	0.007	0.011									
0.780	0.730	0.784	0.723	0.728	0.702	0.713	0.770	0.810	0.728	0.820	0.700	0.713	0.691	0.751	0.807									
0.725	0.862	0.803	0.849	0.766	0.684	0.883	0.788	0.748	0.863	0.736	0.832	0.936	0.944	0.859	0.743									
0.032	0.035	0.034	0.039	0.045	0.035	0.048	0.036	0.034	0.022	0.025	0.034	0.035	0.025	0.027	0.025									
36.9	45.1	40.8	45.2	39.3	35.3	47	41.6	37.7	44.8	37	43.9	48.3	50.1	45.7	36.6									
39.7	38.2	39.9	38.5	37.4	36.2	37.9	40.7	40.9	37.8	41.2	36.9	36.8	36.7	40	39.8									
23.4	16.7	19.3	16.3	23.3	28.4	15.1	17.7	21.4	17.4	21.8	19.1	14.9	13.2	14.3	23.6									

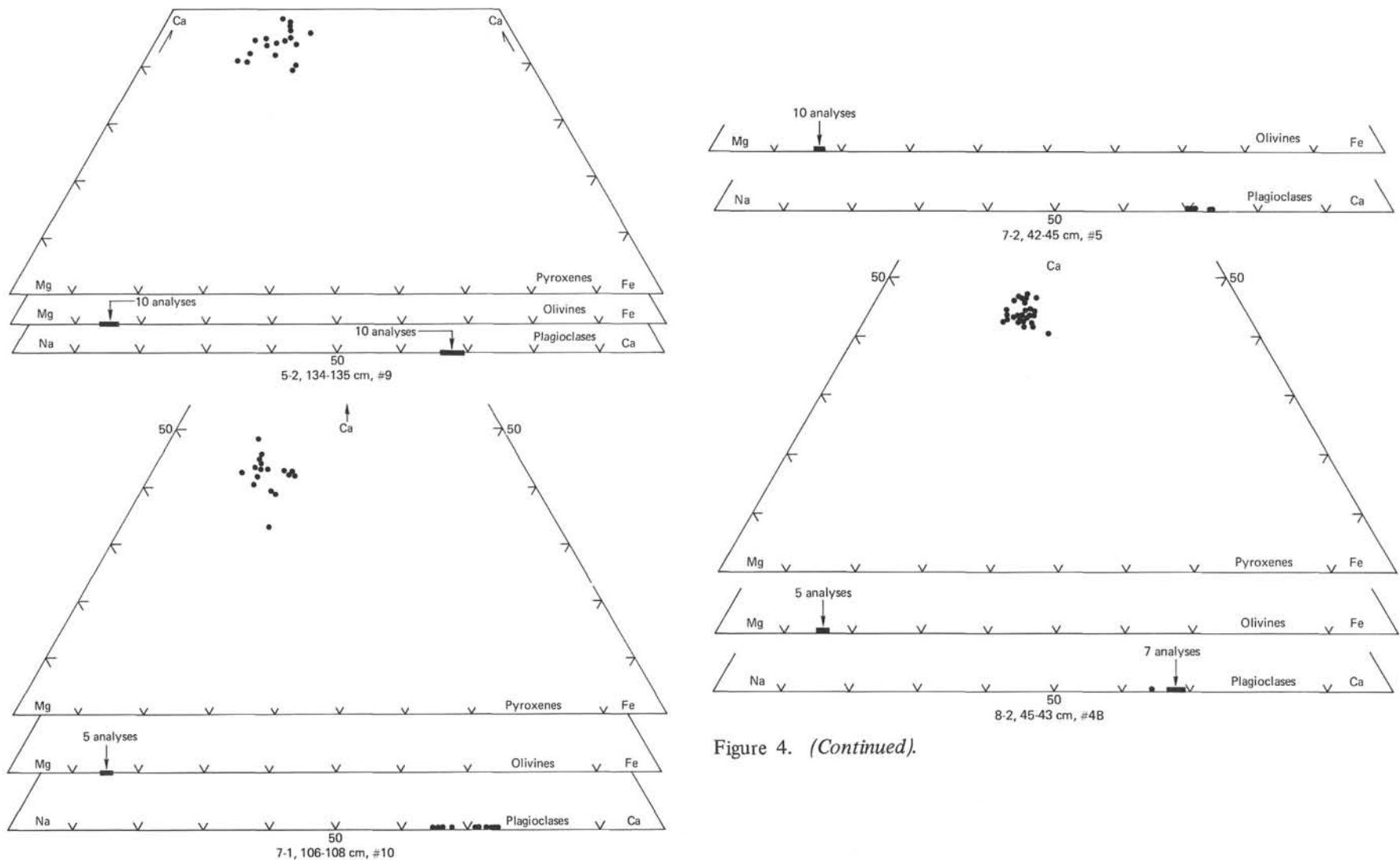


Figure 4. (Continued).

Figure 4. *Ca-Mg-Fe pyroxene compositions, Mg-Fe olivine compositions, and K-Ca-Na plagioclase compositions of the analyzed samples.*

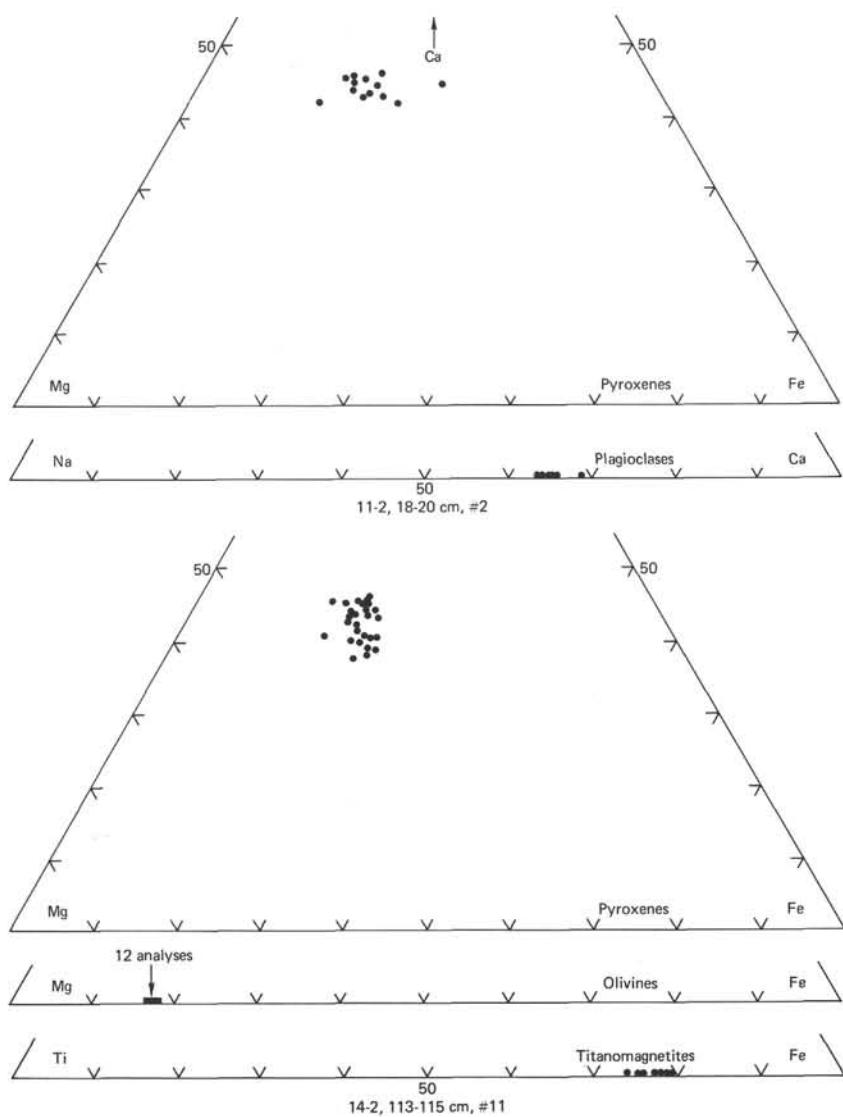


Figure 4. (Continued).

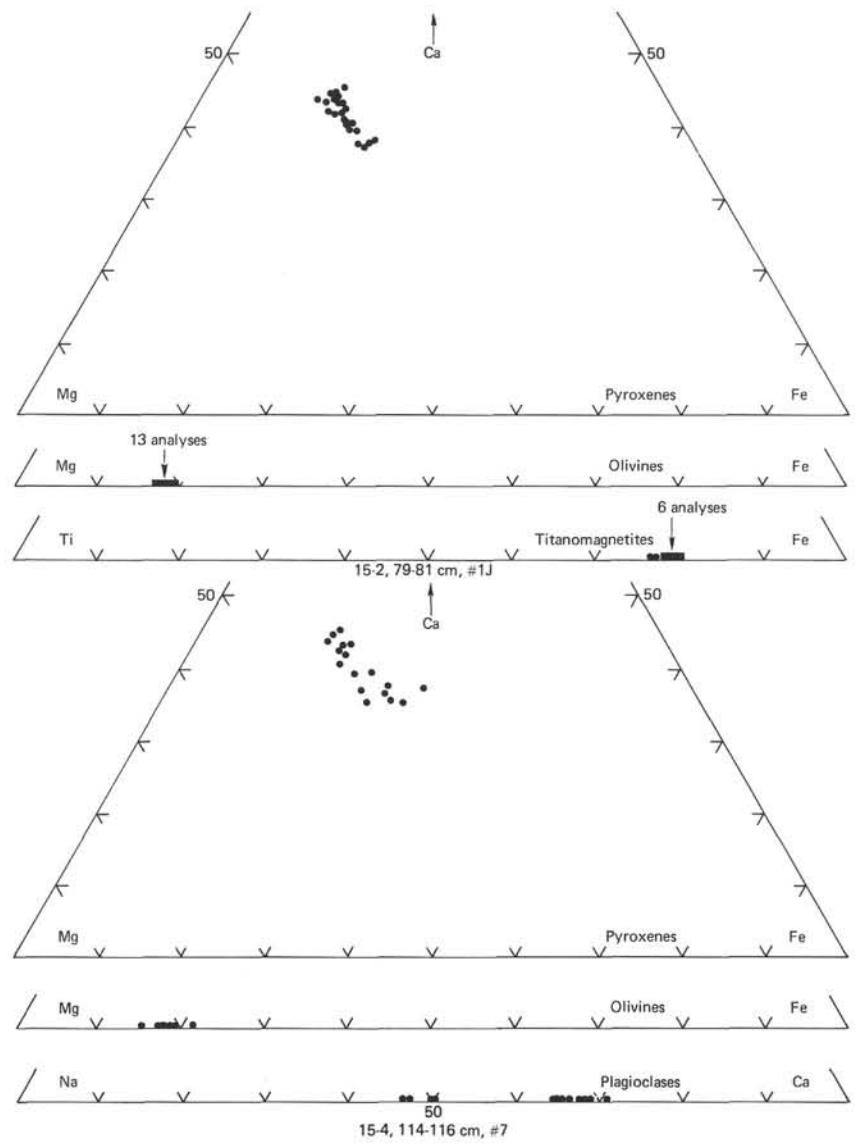


Figure 4. (Continued).

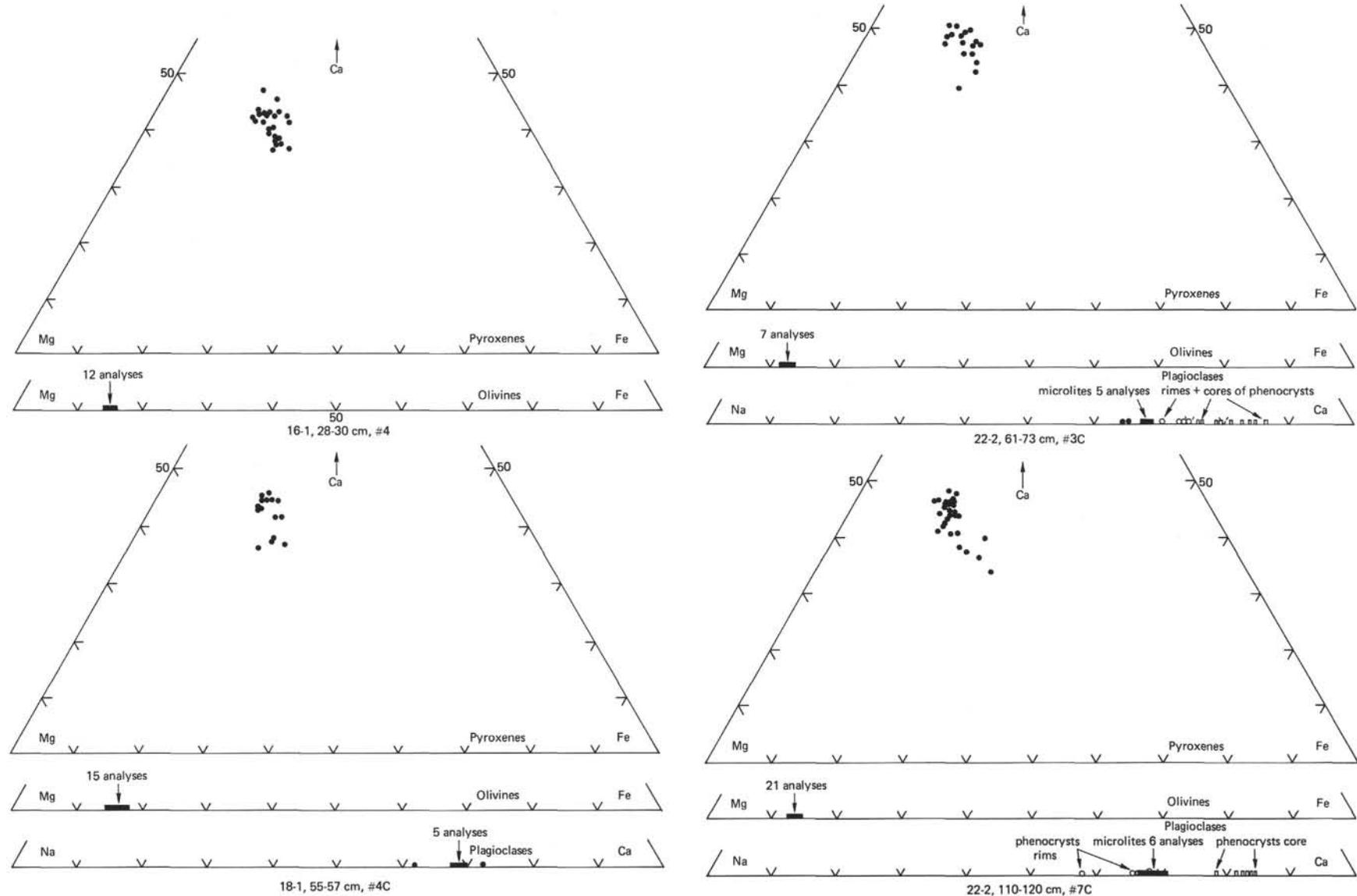


Figure 4. (Continued).

Figure 4. (Continued).

TABLE 5
Representative Plagioclase Analyzed by Microprobe

Sample	5-2, 134-136 cm	7-1, 106-108 cm	7-1, 106-108 cm	7-2, 42-45 cm	8-2, 43-45 cm	11-2, 18-20 cm	15-4, 114-116 cm
SiO ₂	52.26	52.71	53.00	51.88	52.03	52.51	52.94
Al ₂ O ₃	30.42	30.07	30.43	30.75	30.28	29.02	29.99
CaO	14.42	14.80	13.47	14.26	14.15	13.29	13.77
Na ₂ O	3.57	3.05	3.90	3.37	3.63	4.12	3.97
Total	100.67	100.63	100.80	100.26	100.09	98.94	100.68
Si	9.437	9.508	9.527	9.398	9.445	9.626	9.543
Al	6.475	6.394	6.448	6.564	6.840	6.271	6.373
Ca	2.790	2.861	2.595	2.768	2.753	2.611	2.660
Na	1.249	1.065	1.360	1.182	1.277	1.463	1.388
Z	15.92	15.90	15.98	15.97	15.93	15.90	15.92
X	4.04	3.93	3.96	3.95	4.03	4.07	4.05
Ab	30.9	27.1	34.4	29.9	31.7	35.9	34.3
An	69.1	72.9	65.6	70.1	68.3	64.1	65.7

TABLE 5 – *Continued*

Sample	18-1, 55-57 cm		22-2, 61-73 cm		22-2, 110-120 cm		
SiO ₂	51.84	46.34	48.70	52.19	50.76	47.57	52.09
Al ₂ O ₃	30.87	33.43	31.59	30.45	30.39	33.93	29.84
CaO	14.21	17.26	15.22	14.06	13.98	17.36	14.36
Na ₂ O	3.55	1.79	2.96	3.68	3.58	2.00	3.38
Total	100.47	98.82	98.47	100.37	98.71	100.86	99.67
Si	9.376	8.621	9.038	9.444	9.352	8.666	9.492
Al	6.581	7.330	6.910	6.494	6.598	7.285	6.410
Ca	2.754	3.440	3.027	2.727	2.759	3.388	2.804
Na	1.245	0.646	6.910	1.290	1.280	0.706	1.196
Z	15.96	15.95	15.95	15.94	15.95	15.95	15.90
X	4.00	4.09	4.09	4.02	4.04	4.09	4.00
Ab	31.1	15.8	26	32.1	31.7	17.2	29.9
An	68.9	84.2	74	67.9	68.3	82.9	70.1

TABLE 6
Representative Olivines Analyzed by Microprobe

Sample	5-2, 134-136 cm	7-1, 106-108 cm	7-2, 42-45 cm	8-2, 43-45 cm	14-2, 113-115 cm	15-2, 79-81 cm	15-4, 114-116 cm
SiO ₂	39.12	39.50	39.70	38.96	39.54	39.36	39.40
FeO ⁺	14.08	14.38	15.44	15.13	16.21	15.84	18.64
MgO	45.17	46.51	45.32	45.96	44.56	45.07	42.84
Total	98.37	100.39	100.45	100.05	100.30	100.27	100.88
Si	0.995	0.985	0.993	0.961	0.994	0.989	0.996
Fe	0.299	0.300	0.323	0.318	0.341	0.333	0.394
Mg	1.712	1.729	1.690	1.722	1.671	1.688	1.614
Σ	3.006	3.014	3.006	3.026	3.006	3.010	3.004
Fa	14.9	14.8	16	15.8	16.9	16.5	19.6
Fo	85.1	85.2	84	84.2	83.1	83.5	80.4

TABLE 6 – Continued

				Core	Rim		
16-1, 28-30 cm		18-1, 55-57 cm		22-2, 61-73 cm		22-2, 110-120 cm	
40.00	40.70	40.10	38.69	38.62	39.41	38.11	
13.55	14.96	16.91	11.78	12.87	13.62	13.73	
46.89	44.42	43.27	48.69	47.14	47.43	48.93	
100.44	100.08	100.28	99.16	98.64	100.46	100.77	
0.992	1.017	1.010	0.968	0.976	0.975	0.949	
0.281	0.312	0.356	0.247	0.272	0.278	0.286	
1.734	1.654	1.624	1.816	1.776	1.772	1.816	
3.007	2.983	2.990	3.031	3.024	3.025	3.051	
13.9	15.9	18	12	13.3	13.6	13.6	
86.1	84.1	82	88	86.7	86.4	86.4	

TABLE 7
Representative Partial Titanomagnetite
Analyzed by Microprobe

Sample	14-2, 113-115 cm		15-2, 79-81 cm	
FeO ⁺	64.98	63.52	69.50	70.32
TiO ₂	20.54	24.02	21.04	20.39
Total	85.53	87.54	90.10	90.71
Fe	6.375	5.952	6.475	6.572
Ti	1.812	2.024	1.763	1.714
Σ	8.87	7.976	8.238	8.286
(on the basis of 10 oxygens)				