



Table 1. Interstitial water chemistry, Leg 73.

Core-Section (interval in cm)	Sub-bottom depth (m)	pH	Alkalinity (meq-dm <sup>-3</sup> )	S (g/kg)	Ca (mM)	Mg (mM)	Cl (g/kg)	Sr (μM)	Li (μM)	K (mM)	SO <sub>4</sub> (mM)	NH <sub>4</sub> (μM)	Si (μM)
Hole 519													
7-2, 140-150	27-32	7.4	2.95	37.4	10.6	49.9	19.6	—	—	—	—	—	—
17-2, 133-150	66-71	7.4	2.96	35.5	10.8	53.3	20.1	—	—	—	—	—	—
27-2, 112-122	110-115	7.4	3.07	35.5	11.5	51.8	20.1	—	—	—	—	—	—
Hole 520													
3-6, 140-150	108-117	7.43	3.21	35.4	12.5	49.5	19.75	517.3	36.5	10.4	27.0	153.0	407.7
9-4, 140-150	256-266	7.61	5.29	35.2	14.2	45.2	19.17	439.6	43.7	11.2	24.6	229.3	824.5
12-1, 140-150	285-305	7.67	5.28	35.2	15.0	45.7	19.07	368.6	45.5	9.6	25.0	240.3	722.9
16-4, 140-150	313-323	7.70	5.38	35.2	14.9	46.2	19.41	368.6	45.5	9.6	25.0	240.3	722.9
21-2, 94-104	346-352	7.49	4.70	37.4	14.6	47.4	20.39	310.2	35.0	10.6	30.2	0	614.2
Hole 521													
7-2, 140-150	26.30	7.28	2.74	36.0	10.8	52.9	19.80	—	—	—	—	—	—
17-2, 140-150	66-72	7.31	2.73	35.5	10.4	51.4	19.77	—	—	—	—	—	—
Hole 521A													
17-3, 0-9	68-71	7.29	2.58	35.5	10.2	51.3	19.51	—	—	—	—	—	—
Hole 522													
2-2, 140-150	3-7	7.26	2.80	35.5	10.3	52.9	19.78	—	—	—	—	—	—
17-2, 140-150	64-69	7.28	2.52	36.6	10.6	51.3	19.07	—	—	—	—	—	—
22-2, 140-150	86-90	7.21	2.37	38.2	10.4	52.2	19.27	—	—	—	—	—	—
31-2, 140-150	122-126	7.20	2.29	36.6	10.3	52.4	19.21	—	—	—	—	—	—
Hole 523													
8-1, 144-150	30-34	7.34	2.58	38.0	10.7	49.9	19.37	—	—	—	—	—	—
21-2, 144-150	73-77	7.24	2.49	35.5	10.5	53.7	19.61	—	—	—	—	—	—
30-2, 143-150	107-111	7.28	2.29	36.8	10.8	51.5	19.20	—	—	—	—	—	—
Hole 524													
4-3, 140-150	50	7.46	1.94	35.5	9.9	55.5	19.71	—	—	—	—	—	—
11-5, 140-150	131	7.10	0.39	39.3	27.9	38.6	19.64	—	—	—	—	—	—
15-1, 140-150	154	7.48	0.92	36.6	35.8	34.9	19.64	—	—	—	—	—	—
21-5, 140-150	215	7.47	0.81	35.2	47.9	29.6	19.41	—	—	—	—	—	—
25-6, 140-150	246	7.50	1.12	34.1	52.0	27.4	18.70	—	—	—	—	—	—

## Site 524

The chemistry of the interstitial waters at Site 524 indicates steep gradients in calcium and magnesium below a sub-bottom depth of 50 m (Fig. 3). Changes in calcium and magnesium are not linearly correlated. Uptake of magnesium is indicated in the volcanoclastic sediments of Units 2 and 3. The underlying basalts provide a significant source for dissolved calcium. Similarly, large concentration gradients in calcium and magnesium were observed in the Walvis Ridge sites drilled during Leg 74. We intend to carry out further work on the strontium isotopic composition of the pore fluids of Site 524 as well as of those collected during Leg 74. Such data will help to develop a more complete interpretation of the contribution of volcanic matter to reactions involving the exchange of calcium and magnesium between the solid phases and interstitial waters (Hawkesworth and Elderfield, 1978).

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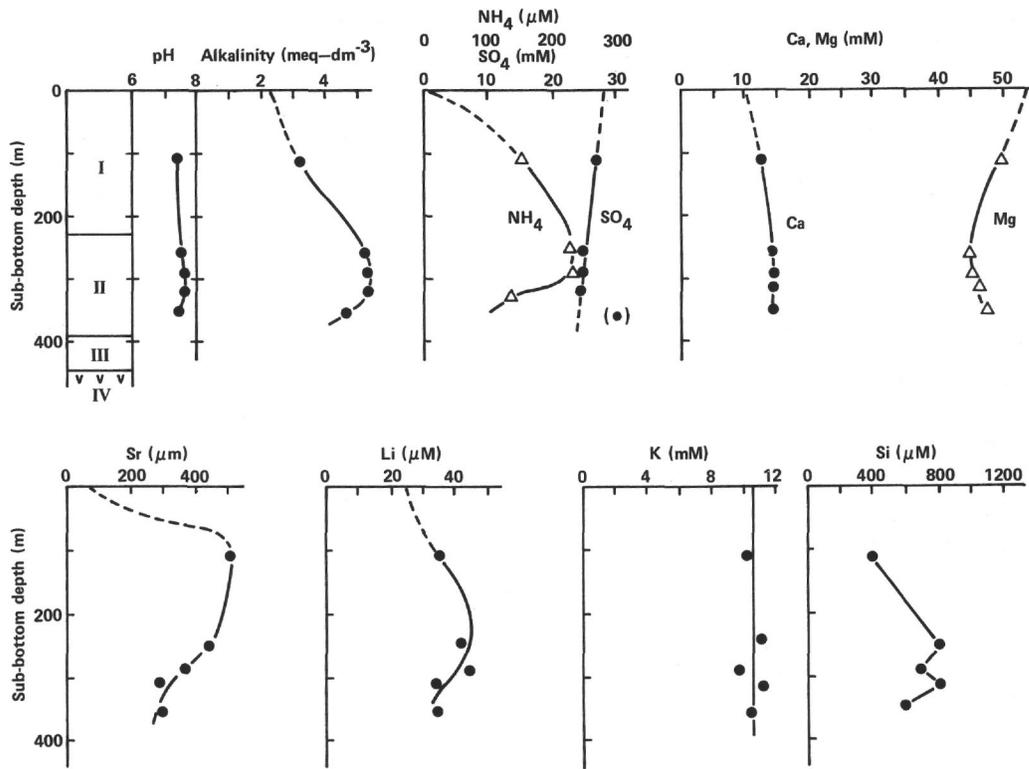


Figure 2. Interstitial water chemistry, Site 520. Lithology is as follows: I = nannofossil ooze; II = nannofossil oozes at 285 m, marls at 300 to 310 m, and diatomaceous oozes at 390 m, respectively; III = nannofossil marl and claystone; IV = basalt.

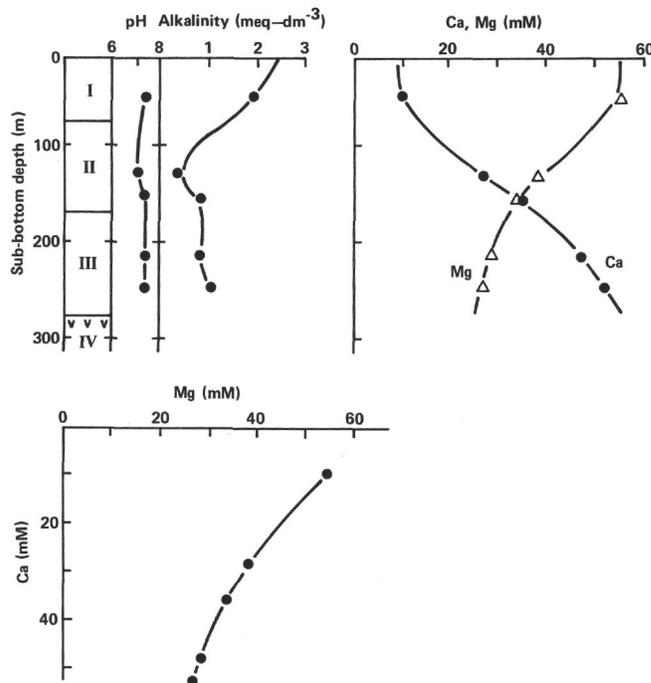


Figure 3. Interstitial water chemistry, Site 524. Lithology is as follows: I = nannofossil ooze and chalk; chert occurs at ~40 m and 72 m; II = nannofossil marls and volcaniclastic sandstones; III = ash, nannofossil claystone, and sandstones; IV = basalts, volcanic breccias, and claystones.